ES106 Video Exercise

Endless Voyage Video Series 7 - Water Properties

Watch the video and answer the following review questions.

- 1. What is the chemical composition of the water molecule?
- 2. What are the three physical states of matter, in which water occurs on planet Earth?
- 3. What is the bond angle between hydrogen and oxygen atoms in a water molecule.
- 4. Draw a sketch of the water molecule, and explain why it is called "dipolar".
- 5. Why do water molecules cluster or clump together?
- 6. True or False: the boiling point of water is unusually low, because of it's chemical composition and dipolar nature.
- 7. True or False: water is not effective at dissolving chemical compounds.
- 8. True or False: many chemical compounds on planet Earth are not soluble (dissolvable) in water.
- 9. Why is it important that water in a solid state, ice, is less dense and floats on liquid water.
- 10. List three types of reservoirs on the Earth surface in which water is located.
- 11. How does ocean water differ chemically from river water?
- 12. List four sources of dissolved elements in seawater that make it "salty".
- 13. True or False: the oceans are saltier than any other body of water on the planet.
- 14. List two mechanisms by which dissolved salts are removed from the ocean, at "salt sinks"

- 15. Define the term "salinity" and explain how it is measured.
- 16. True or False: increasing the amount of dissolved salts in water, increases the electrical conductivity of the water.
- 17. Define the term "density".
- 18. True or False: increasing salinity of seawater results in decreasing density.
- 19. True or False: cold water is more dense than warm water.
- 20. True or False: warm equatorial ocean water rises and cold polar ocean water sinks.
- 21. Explain how heat is transferred from the polar regions to the equatorial regions. Draw a sketch showing generalized circulation of ocean water across the Earth.
- 22. Define the term "thermohaline" circulation and the factors that cause water to move in the oceans.
- 23. True or False: gasses in the atmosphere, such as carbon dioxide, freely exchange with the oceans over time.
- 24. True or False: removal of carbon dioxide from the atmosphere by ocean exchange causes climate temperature to decrease.
- 25. Discuss how the evolution of plants on Earth have impacted its atmospheric gas composition.
- 26. True or False: increase in ocean plant biomass, encourages carbon dioxide consumption from the atmosphere, and forces global atmospheric cooling.
- 27. What chemical element is hypothesized as essential for abundant ocean plant growth.
- 28. Provide a summary statement of what you have learned about the importance of oceans to planet Earth in this video.