Key Do not remove!

GS104 Lab Exercise

Introduction to Scientific Inquiry and Data Analysis

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Science employs the scientific method via qualitative and quantitative observation, the collection of data, which is formulation / testing, and hypothesis modification. This lab exercise provides a basic introduction to establish observation and analysis.

1 - Unit Conversion

(1054 16) (1kg) = 479,1 kg

the attached metric and English measurement unit conversion tables, complete the following conversions, WALL OF YOUR MATH WORK IN THE SPACE PROVIDED.

$$\frac{205}{(12.05)} \text{ cm} \frac{(2.05)}{(12.05)} \frac{1000}{(12.05)} = \frac{2.05}{(12.05)} \frac{2 \times 10^5 \text{ in}}{(12.05)} \frac{1}{(12.05)} \frac{1}{(12.05)} \frac{1}{(12.05)} = 3.16 \text{ mi}$$

$$\frac{1500}{(12.05)} \text{ mm} \frac{(1.50)}{(12.05)} \frac{100000}{(12.05)} = \frac{1500}{1500} \frac{2 \times 10^9 \text{ ft}}{(12.05)} \frac{2.05}{(12.05)} = \frac{378}{787} \frac{787}{100} \frac{1}{(12.05)} = \frac{378}{787} \frac{787}{100} \frac{1}{(12.05)} = \frac{378}{787} \frac{1}{(12.05)} \frac{126}{(12.05)} \frac$$

(7.8 1/2000 ft)=15600 ft.

 $(15600 ft) \left(\frac{1mi}{5280 ft}\right) = 2.95 mi$

.... 2. Selving Equations

The density of a substance is defined by it's mass divided by it's volume. The equation has the following forms

$$D = M/V$$

where D is density in gm/cm³, M = mass in grams, and V is volume in cm³

1. You measure the mass of a substance as 2356 gm. It's volume is 534 cm³, calculate it's density in gm/cm³. SHOW THE FORMULA AND ALL OF YOUR MATH WORK!

$$0 = \frac{2356 \, \text{gm}}{534 \, \text{cm}^3} = 4.4 \, \frac{\text{gm}}{\text{cm}^3}$$

2. The density of a substance is 9.8 gm/cm³. If you had a volume of 3.8 cm³ of the substance, what would be the corresponding mass in grams? Hint: Rearrange the density equation to solve for mass. **SHOW THE FORMULA AND ALL OF YOUR MATH WORK!**

$$V \times D = \frac{m}{V} \times M = VD = (3.8 \text{ g/s}) (9.8 \text{ g/s}) = 37.2 \text{ g/s}$$

3. The density of a substance is 2.5 gm/cm³ and you possess 15.3 grams of that material. What will be its corresponding volume in cm³. Hint: Rearrange the density equation to solve for mass. SHOW THE FORMULA AND ALL OF YOUR MATH WORK!

$$\frac{1}{0} \times \sqrt{8} = \frac{m}{V} \times \sqrt{\frac{1}{0}} \implies \sqrt{\frac{m}{0}} = \frac{15.3 \text{ g}}{2.5 \text{ g/cm}^3} = 6.1 \text{ cm}^3$$

The velocity of moving objects (for example your car while driving) is measure as a rate of motion, according to the following equation:

$$V = d/t$$

where V is velocity (m/sec), d is distance (m), and t is time (sec).

4. You drive your car between two cities that are 123 miles apart. It takes you 4 hours to get there. Calculate your average velocity in mi/hr. SHOW THE FORMULA AND ALL OF YOUR MATH WORK!

$$V = \frac{d}{t} = \frac{123 \text{ ni}}{4 \text{ hr}} = 30.75 \frac{\text{mi}}{\text{hr}}$$

5. Using the velocity you caculated in 4 above, what was your velocity in m/sec? Hint: you will have to use a distance and time conversion factor. SHOW THE FORMULA AND ALL OF YOUR MATH WORK!

6. You are driving a car at a velocity of 10 m/sec for a distance of 12 km. How long did it take you to get there? Answer in hours. SHOW THE FORMULA AND ALL OF YOUR MATH WORK!

$$\frac{V=d}{t} \qquad (12 \text{ k/s}) \left(\frac{1000 \text{ m}}{1 \text{ k/s}}\right) = 12000 \text{ m} \qquad 1200 \text{ f} \left(\frac{1 \text{ hr}}{3600 \text{ s}}\right) = 0.73 \text{ hr}$$

$$2 \quad t = \frac{d}{V} = \frac{12000 \text{ m}}{10 \text{ m}} = 12000 \text{ s}$$

$$1200 \text{ f} \left(\frac{1 \text{ hr}}{3600 \text{ s}}\right) = 0.73 \text{ hr}$$

$$2 \quad t = \frac{d}{V} = \frac{12000 \text{ m}}{10 \text{ m}} = 12000 \text{ s}$$