Glycolysis

- Introduction: Glucose utilization
- Glycolysis
- Entry of glucose into the cell
- Preparatory phase of glycolysis
- Energy production

Glycolysis Embden-Meyerhoff pathway

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Glycolysis

- Ancient Pathway
- In cytoplasm
- No oxygen required
- Used for energy production
- Production of intermediates for other pathways
- Found in tissues with limited blood supply

Entry of glucose into the cell

- Transport
- hexokinase
- glucokinase in liver
- $\bullet\,$ hexokinase vs glucokinase
- forms anion to keep in cell

Glucose Transporters

Kinetic properties of hexokinase

Preparatory phase of glycolysis

- •2 ATP
- •Phosphofructokinase (PFK-1)
- •regulated
- •allosterically

Aldolase

Mechanism: Aldolase

- **Nucleophillic (electron rich group attacks electron deficient nucleus attack on #2 carbon.**
- • **Lysine residue forms a Schiff base**
- **OH (from C) and H from cysteine form H ²O.**
- **Thiolate gets H + from substrate, release Gly 3-P**
- **His residue donates H+**
- • **Hydrolysis of Schiff base**

Energy production

- 1,3 B P G A
- \bullet PEP
- •4 ATP & 2 NADH
- Pyruvate end product

Phosphoglycerate Kinase

1-Arseno-3-phosphoglycerate

3-Phosphoglycerate

(Figure 12.14). This nonenzymatic hydrolysis produces 3-phosphoglycerate and regenerates inorganic arsenate, which can again react with a thioacyl-enzyme intermediate. Glycolysis can proceed from 3-phosphoglycerate, but the ATP-producing reaction involving 1,3-bisphosphoglycerate is bypassed. As a result, there is no net formation of ATP from glycolysis, with potentially lethal consequences.

Mechanism: dehydrogenase

- • **Ionized cysteine attacks C-1, forming thiohemiacetal (aldehyde/thio group)**
- \bullet **Hydride ion (H-) reduces NAD +, forms thioacyl intermediate**
- **Phosphate enters displaces thioacyl**
- **1,3 Bisphsopho disassociates.**

PEP to Pyruvate

Biological Systems

- **Net 2 ATP**
- **2 NADH**
- **Most reactions at equilibrium can be reversed**

Table 15-3

Typical concentrations of glycolytic intermediates in erythrocytes

After S. Minakami and H. Yoshikawa. Biochem. Biophys. Res. Comm. 18(1965):345.

Overall reactions of glycolysis

TABLE 14-2 Free-Energy Changes of Glycolytic Reactions in Erythrocytes

Note: $\Delta G^{\prime\circ}$ is the standard free-energy change, as defined in Chapter 13 (p. 491). ΔG is the free-energy change calculated from the actual concentrations of glycolytic intermediates present under physiological conditions in erythrocytes, at pH 7. The glycolytic reactions bypassed in gluconeogenesis are shown in red. Biochemical equations are not necessarily balanced for H or charge (p. 506).

Regulation of Glycolysis

- ATP/AMP ratios are important
- Two roles: energy production and building blocks for biosynthesis

Regulation of Hexokinase

Phosphofructokinase: Highly regulated

- Allosteric enzyme:
- Activated by ADP and AMP
- Inhibited by ATP and Citrate (from TCA cycle)
- Fructose 2,6 bisphosphate regulation

PFK-2

- **Tandem enzyme: PFK-2, both kinase and phsophotase together**
- **Fru 2,6 P potent activator of PFK-1**
- • **prevents inhibition of citrate/ATP for fatty acid biosynthesis**
- • **Relative velocity curves for PFK-1**
- • **Effect of glucagon on PFK-1**

Activation of PFK-1 by Fru 2,6 Bis

Other sites of Regulation

- **Pyruvate kinase**
- •**allosteric**
- • **stimulated by fructose 1,6P**
- **inhibited by acetyl-CoA, fatty acids**
- • **protein kinase inhibits pyruvate kinase**
- **Glyceraldehyde 3-P dehydrogenase**
- **stimulated by NAD+**

